

اللَّهُ نُورُ السَّمَاوَاتِ وَالْأَرْضِ



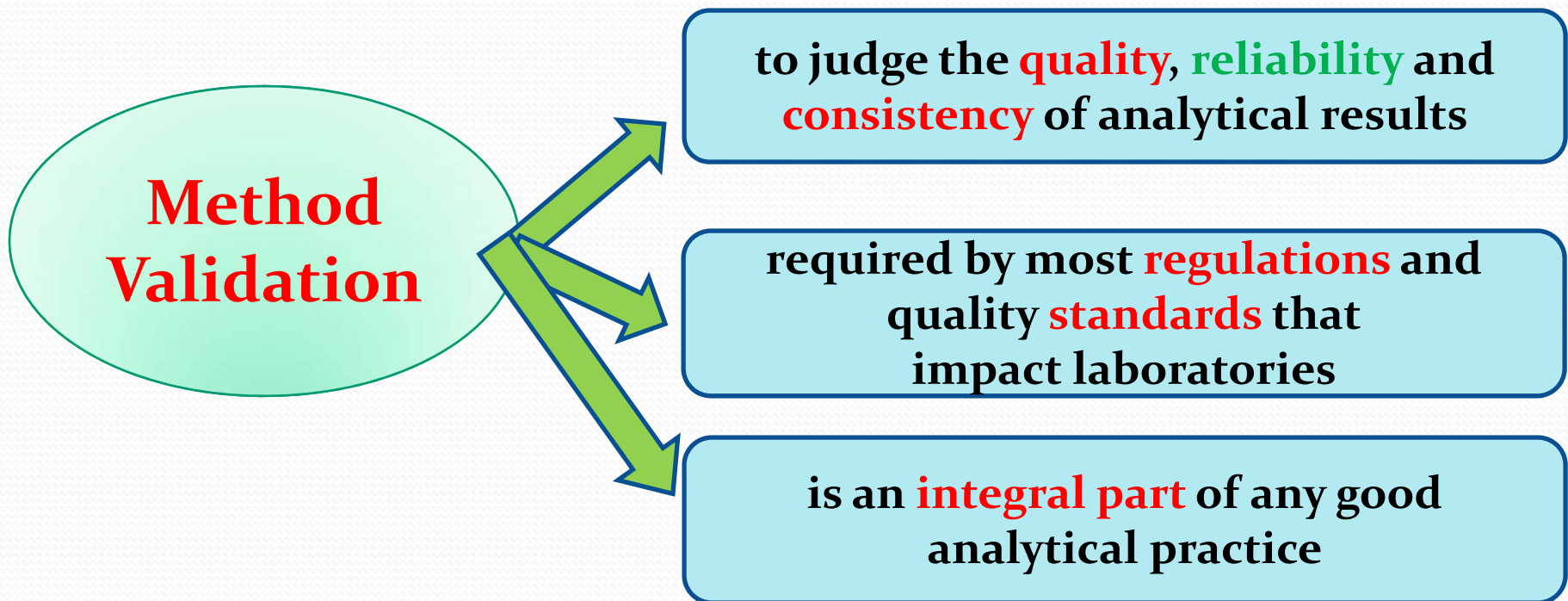
Validation of Analytical Methods



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Why Method Validation?

- **Objective** of any analytical measurement is:
to obtain **consistent, reliable** and **accurate data**



When Method Validation?

Analytical methods need to be **validated**, **verified**, or **revalidated** in the following instances:

- Before **initial use** in routine testing
- When transferred to **another laboratory**
- Whenever the **conditions or method parameters** for which the method has been validated **change** (for example, an **instrument** with different characteristics or samples with a different **matrix**) and the change is outside the original scope of the method.

International regulatory bodies and their guidelines

<i>Body</i>	<i>Full name</i>	<i>Guidance on</i>
<i>Eurachem</i>	<i>Focus for Analytical Chemistry in Europe</i>	<i>Method validation</i>
<i>CITAC</i>	<i>Cooperation of International Traceability in Analytical Chemistry</i>	<i>Proficiency testing Quality Assurance</i>
<i>EA</i>	<i>European Cooperation for Accreditation</i>	<i>Accreditation</i>
<i>CEN</i>	<i>European Committee for Normalization</i>	<i>Standardization</i>
<i>IUPAC</i>	<i>International Union of Pure & Applied Chem.</i>	<i>Method validation</i>
<i>ISO</i>	<i>International Standardization Organisation</i>	<i>Standardisation</i>
<i>AOAC</i>	<i>Association of Official Analytical Chemists</i>	<i>Internal qual. Control Proficiency testing</i>
<i>ILAC</i>	<i>International Laboratory Accreditation Cooperat.</i>	<i>Accreditation</i>
<i>FDA</i>	<i>US Food and Drug Administration</i>	<i>Method validation</i>
<i>USP</i>	<i>United States Pharmacopoeia</i>	<i>Method validation</i>
<i>ICH</i>	<i>International Conference on Harmonization</i>	<i>Method validation</i>



A Unique Approach

- **International Conference on Harmonisation (ICH)** was created in 1990
- Agreement between the **EU**, **Japan** and the **USA** to harmonize different regional requirements for registration of pharmaceutical **drug** products
- **Unique** because joint effort by regulators and associated **pharmaceutical** industry trade associations



What is Validation?

- ICH Guideline **Q₂(R₁)** – Text on Validation of Analytical Procedures

“The **objective** of validation of an analytical procedure is to demonstrate that it is **suitable** for its **intended** purpose.”

Examples of Methods That Require Validation Documentation

- **Chromatographic Methods - HPLC, GC, TLC, GC/MS, etc.**
 - **Pharmaceutical Analysis**
 - **Bioanalytical Analysis - Clinical Studies.**
- **Spectrophotometric Methods – UV/VIS, IR, NIR, AA, NMR**
- **Capillary Electrophoresis Methods**
- **Particle Size Analysis Methods - Laser, Microscopic**
- **Automated Analytical Methods**

Considerations Prior to Method Validation

- **Suitability of Instrument**
 - Status of Qualification and Calibration
- **Suitability of Materials**
 - Status of Reference Standards, Reagents, Placebo Lots
- **Suitability of Analyst**
 - Status of Training and Qualification Records
- **Suitability of Documentation**
 - Written analytical procedure and proper approved protocol with pre-established acceptance criteria

Types of Analytical Procedures to be Validated

- Identification tests
- Quantitative tests
- Limit tests for the control of impurities;
- Quantitative tests of the active moiety in samples of drug substance or drug product or other selected component(s) in the drug product.



Typical validation characteristics which should be considered

- Accuracy
- Precision
 - Repeatability
 - Intermediate Precision
- Specificity
- Detection Limit
- Quantitation Limit
- Linearity
- Range

ICH Validation Characteristics vs. Type of Analytical Procedure

Type of Analytical Procedure	Identification	Impurity testing		Assay
		Quantitative	Limit Tests	
Accuracy	No	Yes	No	Yes
Precision				
Repeatability	No	Yes	No	Yes
Interm. Prec.	No	Yes	No	Yes
Specificity	Yes	Yes	Yes	Yes
LOD	No	No	Yes	No
LOQ	No	Yes	No	No
Linearity	No	Yes	No	Yes
Range	No	Yes	No	Yes

Specificity

Specificity is the **ability** to assess **unequivocally** the **analyte** in the presence of components which may be expected to be present. Typically these might include impurities, degradants, matrix, etc.

On the other hand:

Degree to which the assay distinguishes between **the target analyte** and other **components** in the sample matrix;

- **the higher the analytical specificity, the lower the level of false-positives.**



Specificity

- **Note:**

- **Lack** of **specificity** of an individual analytical procedure may be compensated by **other supporting analytical procedure(s)**
- Other international authorities (**IUPAC, AOAC**) have preferred the term “**Selectivity**” reserving “**Specificity**” for those procedures that are completely selective

Specificity

- Some analytical procedures are **not sufficiently specific** for the intended purpose
 - Assay by **titration**
 - Assay of **enantiomer** by **achiral** method
 - Identification by **UV absorbance**

Linearity

- The **linearity** of an analytical procedure is its **ability** (within a given range) to obtain **test results** which are directly proportional to **the concentration** (amount) of **analyte** in the sample.
- A **calibration** (standard) **curve** is the relationship between **instrument response** and known **concentrations of the analyte**.

Linearity Should be Evaluated

- **By Visual Inspection of plot of signals vs. analyte concentration**
- **By Appropriate statistical methods**
 - **Linear Regression ($y = mx + b$)**
 - **Correlation Coefficient, y-intercept (b), slope (m)**
- **Acceptance criteria: Linear regression $r^2 > 0.95$**

Requires a minimum of 5 concentration levels



Range

- The range of an analytical procedure is the **interval** between the **upper** and **lower concentration** (amounts) of **analyte** in the sample (including these concentrations) for which it has been demonstrated that the analytical procedure has a **suitable level of precision, accuracy and linearity**.
- The specified range is normally **derived from linearity** studies and depends on the **intended** application of the procedure.



Accuracy

- The **accuracy** of an analytical procedure expresses the **closeness** of agreement between the **value** which is accepted either as a **conventional true** value or an **accepted reference** value and the value found.
- Note:
This is sometimes termed **trueness**.



Accuracy

Recommended Data

- **Accuracy** should be assessed using a minimum of 9 determinations over a minimum of **3 concentration** levels covering the specified range (e.g., 3 concentrations/3 replicates each of the total analytical procedure).
- Accuracy should be reported as **percent recovery** by the assay of known **added amount** of **analyte** in the sample

Accuracy Data

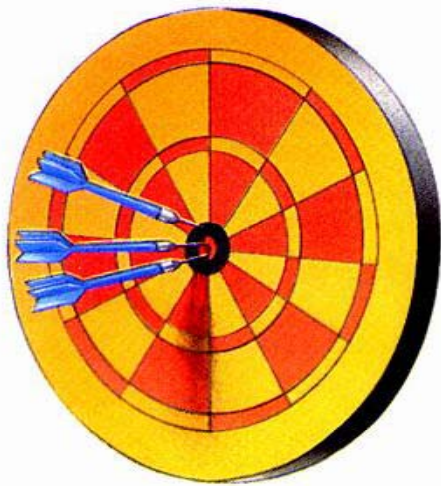
Amount Added (mg)	Amount Found (mg)	Percent Recovery
0.0	0.0	---
50.2	50.4	100.5
79.6	80.1	100.6
99.9	100.7	100.8
120.2	119.8	99.7
150.4	149.7	99.5



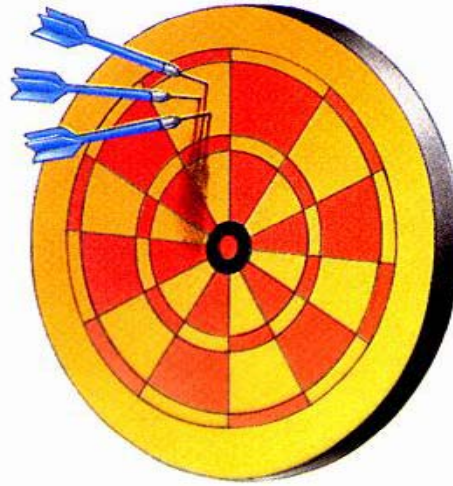
Precision

- The **precision** of an analytical procedure expresses the **closeness** of **agreement** (degree of scatter) between a **series of measurements** obtained from **multiple sampling** of the **same** homogeneous **sample** under the prescribed conditions.

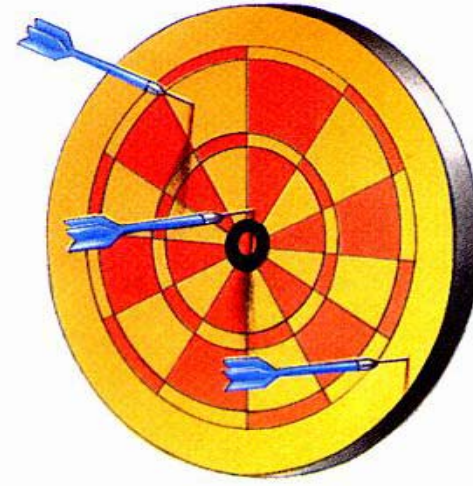
Accuracy v.s Precision



Good accuracy
Good precision



Poor accuracy
Good precision



Poor accuracy
Poor precision

Precision

Precision may be considered at three levels:

- **Repeatability**

Repeatability expresses the precision **under the same operating conditions over a short interval of time**. Repeatability is also termed **intra-assay precision**.

- **Intermediate precision**

Intermediate precision expresses **within-laboratories** variations: **different days, different analysts, different equipment**, etc.

- **Reproducibility**

Reproducibility expresses the **precision** between **laboratories** (**collaborative studies**, usually applied to **standardization of methodology**).



Precision

- Precision should be investigated using **homogeneous, authentic samples**.
- The precision of an analytical procedure is usually expressed as the **variance, standard deviation** or **coefficient of variation** of a series of measurements.

Repeatability & Intermediate Precision

Day 1	Day 2	Day 2
101	100	98
101	103	102
100	101	98
100	99	97
101	100	101
103	100	102

Mean = 100.9
RSD = 1.5%

Mean = 100.3
RSD = 0.51%

Mean = 99.7
RSD = 1.6%

Grand
Mean = 100.3
RSD = 1.63%

Reproducibility

Lab 1		Lab 2		Lab 3	
Day 1	Day 2	Day 1	Day 2	Day 1	Day 2
Man 1	Man 2	Man 1	Man 2	Man 1	Man 2
3 Prep	3 Prep	3 Prep	3 Prep	3 Prep	3 Prep

Detection limit (DL)

- The **detection limit** of an individual analytical procedure is the **lowest amount of analyte** in a sample which can be detected but **not necessarily quantitated** as an exact value.
- Note :
- Detection limit (DL) = Limit of Detection
- Synonymous with “ **Sensitivity**”



Quantitation limit (QL)

- The **quantitation limit** of an individual analytical procedure is **the lowest amount of analyte** in a sample which can be **quantitatively** determined with **suitable precision and accuracy**.
- **Note:**
- Quantitation limit (QL)= Limit of Quantitation (LOQ)

LOD and LOQ Estimated by

1. Based in **Visual** Evaluations
 - Used for non-instrumental methods
2. Based on **Signal-to Noise-Ratio**
 - 3:1 for Detection Limit
 - 10:1 for Quantitation Limit
3. Based on **Standard Deviation** of the Response and the **Slope**

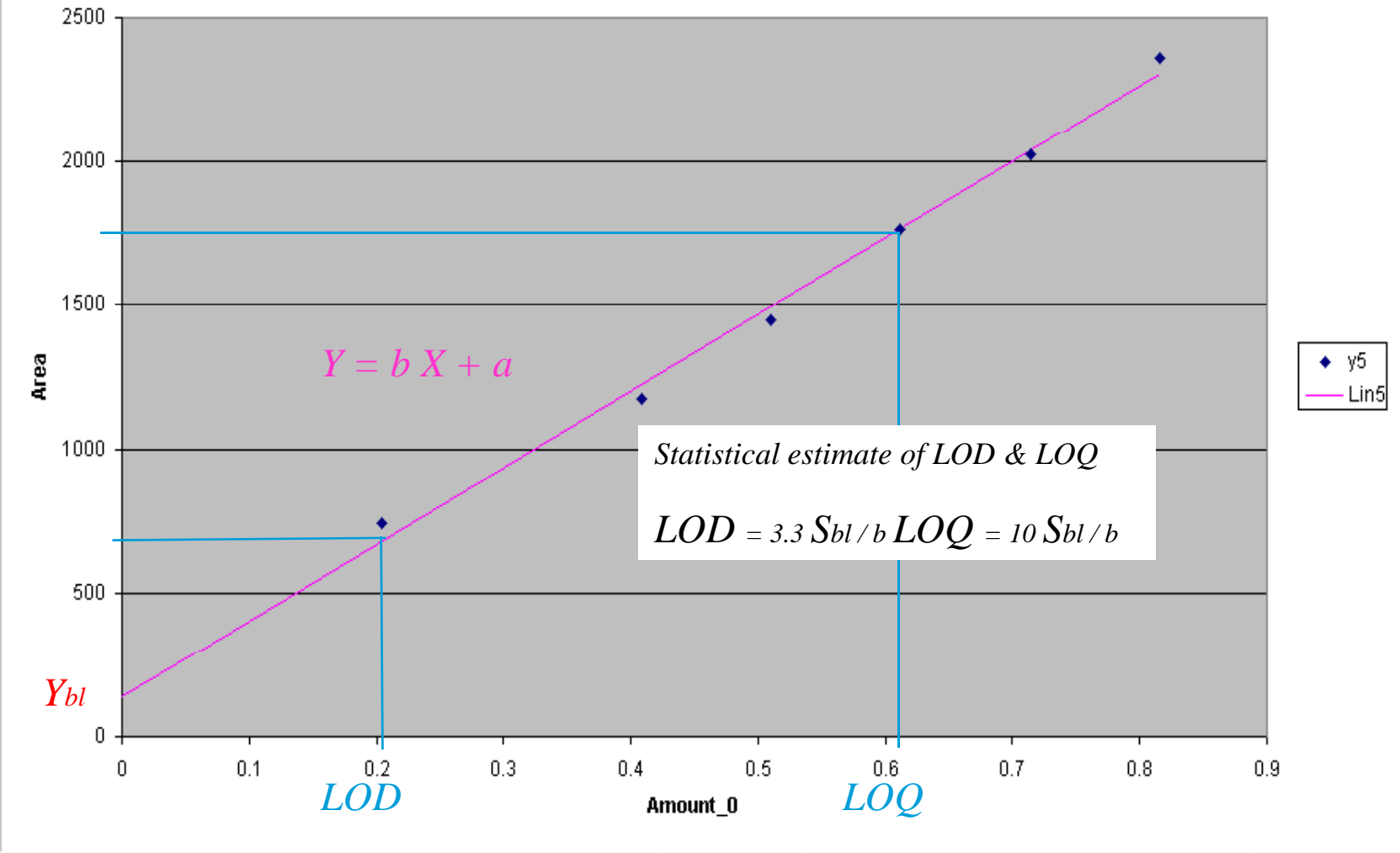
LOD and LOQ Estimated by

$$\text{LOD(DL)} = \frac{3.3s}{b}$$

$$\text{LOQ(QL)} = \frac{10s}{b}$$

- **b = slope of calibration curve**
- **s = standard deviation of regression line**

Limit of Detection



Robustness

- **Definition:** The robustness of an analytical procedure is a measure of **its capacity** to remain **unaffected** by small, but **deliberate variations** in method parameters and provides an indication of its **reliability** during normal usage.
- **Determination:** Comparison results under **differing conditions** with precision under normal conditions
- **Examples of typical variations in LC**
 - Influence of variations of pH in a mobile phase
 - Influence of variations in mobile phase composition
 - Different columns (different lots and/or suppliers)
 - Temperature
 - Flow rate

Ruggedness

- **Degree of reproducibility** of test results under a variety of conditions
 - Different Laboratories
 - Different Analysts
 - Different Instruments
 - Different Reagents
 - Different Days
 - Etc.
- Expressed as %RSD

با تشکر فراوان

